



Course E-Syllabus

1	Course title	Special topics in analytical chemistry		
2	Course number	0353411		
3	Credit hours	3 hours		
	Contact hours (theory, practical)	1:15 hour everyday		
4	Prerequisites/corequisites	0333312		
5	Program title	Bachelor in Chemistry		
6	Program code	03		
7	Awarding institution	The University of Jordan		
8	School	Faculty of Science		
9	Department	Department of Chemistry		
10	Level of course	Final year		
11	Year of study and semester (s)	Summer semester 2019/2020		
12	Final Qualification	According to University regulations		
13	Other department (s) involved in teaching the course	None		
14	Language of Instruction	English		
15	Teaching methodology	□Blended ⊠Online		
16	Electronic platform(s)	⊠Moodle ⊠Microsoft Teams □Skype □Zoom □Others		
17	Date of production/revision	13/7/2020		
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19 Other instructors:

ame:	
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20 Course Description:

As stated in the approved study plan. This course covers different analytical techniques that used in research, labor	ratories and industrials.
1 Course aims and outcomes:	
A- Aims:	
The aim of this course is to provide the students with good knowledge on design starting from collecting a sample, followed by selecting the suitable sample preparation Sample preparation such as LLE, SPE, SCF, and many of microextraction techniques	on and analysis technique.
The principle of chromatographic techniques such as high performance liquid chromatograph (GC) using many type of detectors will be also taught. The underst	tanding and differentiation
between different chromatographic mechanisms such as partitioning, ion ex	change and adsorption
chromatography are other objectives of this course.	
B- Intended Learning Outcomes (ILOs): Upon successful completion of this course, students will be able to:	
B1. To design analytical experiment by selecting the suitable extraction and an B2. Understand the differences between different sample preparation technique B3. Understand the differences between different chromatographic techniques.	es.

22. Topic Outline and Schedule:

Week	Lecture	Торіс	Teaching Methods*/platform	Evaluation Methods**	References
	1.1	Chapter 1: Introduction			
	1.2 to analytical method				
1	1.3	Chantar 2: Introduction			
	1.4	Chapter 2: Introduction to sample preparation			
	1.5				
	2.1	Chapter 3: Application			
	2.2	to LLE			
2	2.3	10 222			
	2.4				
	2.5	Chapter 4: Application			
	3.1	to SPE			
	3.2				Handbook of
3	3.3	Chantar 5: Sunnar	Synchronous		instrumental
	Chapter 5: Supper critical fluid extraction	lecturing/meeting		techniques For analytical	
	3.5	orrada fraid extraction	/Microsoft Teams	Quizzes,	chemistry (Frank Settle, editor)
	4.1	Chantan (assignments and Final exam	
	4.2	Chapter 6: Microextraction		Filial Exam	
4	4.3	Whereextraction			
	4.4				
	4.5	Chapter 7: Liquid phase			
	5.1	microextraction			
	4.2				
5	5.3	C1			
	5.4	Chapter 8: Application to microextraction			
	5.5	to inicroextraction			
	6.1				
	6.2 Chapter 9: Introduction	to chromatography			
6	6.3	to emomatography			
	6.4				
	6.5	Chapter 10: Gas chromatography			
	7.1	Cinomatography			
	7.2				
7	7.3	Chapter 11: High			
	7.4	performance liquid chromatography			
	7.5	- Cinomatography			
	8.1		1		1
	8.2	Presentations			
8	8.3			Onel massautation	
	8.4			Oral presentation	
	8.5	1		1	

- Teaching methods include: Synchronous lecturing/meeting; Asynchronous lecturing/meeting
- Evaluation methods include: Homework, Quiz, Exam, pre-lab quiz...etc

23 Evaluation Methods:

Opportunities to demonstrate achievement of the ILOs are provided through the following assessment methods and requirements:

Evaluation Activity	Mark	Topic(s)	Period (Week)	Platform
Assignment	10	Ch.2+3	3	Moodle
Quiz	10	Ch4+5+6	5	Moodle
Assignment	10	Ch7+8	6	Moodle
Presentation	20		8	Microsoft teams
Final exam	50	All chapters		In the campus

24 Course Requirements (e.g. students should have a computer, internet connection, webcam, account on a specific software/platform...etc):

Computer and internet connection	

25 Course Policies:

- A- Attendance policies: All students are expected to follow the rules at The University of Jordan. Unexcused absences exceeding 15% of the total lectures will result in "F" grade
- B- Absences from exams and submitting assignments on time: University regulations.
- C- Health and safety procedures: NA
- D- Honesty policy regarding cheating, plagiarism, misbehavior: University regulations
- E- Grading policy: University regulations
- F- Available university services that support achievement in the course: NA

26 References:

A- Required book(s), assigned reading and audio-visuals:

	Uploaded at E-learning
	B- Recommended books, materials and media:
	Handbook of instrumental techniques For analytical chemistry (Frank Settle, editor) Media are uploaded at E-learning.
2	Additional information:
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N	ame of Course Coordinator: Ahmad MakahlehSignature: Date: -13/7/2020
F	ead of Curriculum Committee/Department: Signature:
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H	ead of Curriculum Committee/Faculty: Signature:
Γ	ean: Signature: